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Form PTO-1390		U.S. DEPARTMENT OF COMMERCE PATENT AND TRADEMARK OFFICE	ATTORNEY'S DOCKET NUMBER P22180
TRANSMITTAL LETTER TO THE UNITED STATES DESIGNATED/ELECTED OFFICE (DO/EO/US) CONCERNING A FILING UNDER 35 U.S.C. 371			U.S. APPLICATION NO. (If known, see 37 CFR 1.5) Not Yet Assigned <b>10/088531</b>
INTERNATIONAL APPLICATION NO. PCT/JP00/06951	INTERNATIONAL FILING DATE 5 October 2000	PRIORITY DATE CLAIMED 7 October 1999	
TITLE OF INVENTION: METHOD OF TREATING EPOXY RESIN-CURED PRODUCT			
APPLICANT(S) FOR DO/EO/US: Katsuji SHIBATA, Hiroshi SHIMIZU, Ayako MATSUO, and Takeshi HORIUCHI			
Applicant herewith submits to the United States Designated/Elected Office (DO/EO/US) the following items and other information.			
<p>1. <input checked="" type="checkbox"/> This is a FIRST submission of items concerning a filing under 35 U.S.C. 371.</p> <p>2. <input type="checkbox"/> This is a SECOND or SUBSEQUENT submission of items concerning a filing under 35 U.S.C. 371.</p> <p>3. <input type="checkbox"/> This is an express request to promptly begin national examination procedures (35 U.S.C. 371(f)).</p> <p>4. <input checked="" type="checkbox"/> The US has been elected by the expiration of 19 months from the priority date (PCT Article 31).</p> <p>5. <input type="checkbox"/> A copy of the International Application as filed (35 U.S.C. 371(c)(2))</p> <p style="margin-left: 20px;">a. <input type="checkbox"/> is attached hereto (required only if not communicated by the International Bureau).</p> <p style="margin-left: 20px;">b. <input checked="" type="checkbox"/> has been communicated by the International Bureau.</p> <p style="margin-left: 20px;">c. <input type="checkbox"/> is not required, as the application was filed in the United States Receiving Office (RO/US).</p> <p>6. <input checked="" type="checkbox"/> An English language translation of the International Application as filed (35 U.S.C. 371 (c)(2)).</p> <p>7. <input type="checkbox"/> Amendments to the claims of the International Application under PCT Article 19 (35 U.S.C. 371(c)(3))</p> <p style="margin-left: 20px;">a. <input type="checkbox"/> are attached hereto (required only if not communicated by the International Bureau).</p> <p style="margin-left: 20px;">b. <input type="checkbox"/> have been communicated by the International Bureau.</p> <p style="margin-left: 20px;">c. <input type="checkbox"/> have not been made; however, the time limit for making such amendments has NOT expired.</p> <p style="margin-left: 20px;">d. <input type="checkbox"/> have not been made and will not be made.</p> <p>8. <input type="checkbox"/> An English language translation of the amendments to the claims under PCT Article 19 (35 U.S.C. 371(c)(3))</p> <p>9. <input checked="" type="checkbox"/> An oath or declaration of the inventor(s) (35 U.S.C. 371(c)(4)). "Unexecuted"</p> <p>10. <input checked="" type="checkbox"/> An English language translation of the annexes to the International Preliminary Examination Report under PCT Article 36 (U.S.C. 371(c)(5)).</p>			
<b>Items 11 to 16 below concern other document(s) or information included:</b>			
11. Assignee: <u>Hitachi Chemical Co., Ltd., of Tokyo, Japan</u>			
12. <input type="checkbox"/> An Information Disclosure Statement under 37 CFR 1.97 and 1.98.			
13. <input type="checkbox"/> An assignment document for recording. A separate cover sheet in compliance with 37 CFR 3.28 and 3.31 is included			
14. <input type="checkbox"/> A FIRST preliminary amendment.			
<input type="checkbox"/> A SECOND or SUBSEQUENT preliminary amendment.			
15. <input type="checkbox"/> A substitute specification.			
16. <input type="checkbox"/> A change of power of attorney and/or address letter.			
17. <input type="checkbox"/> Figure of Drawing to be published			
18. <input checked="" type="checkbox"/> Other items or information:			
International Application as published (cover page).			
Cover Letter Under 35 U.S.C. 371 and 37 C.F.R. 1.495.			
PCT/IPEA/409 International Preliminary Examination Report.			
PCT/IB/308.			
PCT/IB/304			
PCT/ISA/210.			
Claim of Priority.			

U.S. APPLICATION NO. (If known, see 37 CFR 1.5) Not Yet Assigned		INTERNATIONAL APPLICATION NO. PCT/JP00/06951		ATTORNEY'S DOCKET NUMBER P22180	
19. The following fees are submitted:  Basic National Fee (37 CFR 1.492(a)(1)-(5)): Search report has been prepared by the EPO or JPO. . . . . \$ 890.00 International preliminary examination fee paid to USPTO (37 CFR 1.482). . . . . \$ 710 00 No international preliminary examination fee paid to USPTO (37 CFR 1.482) but international search fee paid to USPTO(37 CFR 1.445(a)(2)). . . . . \$ 740.00 Neither international preliminary examination fee (37 CFR 1.482) nor international search fee (37 CFR 1.445(a)(2) paid to USPTO. . . . . \$1,040.00 International preliminary examination fee paid to USPTO (37 CFR 1.482) and all claims satisfied provisions of PCT Article 33(2)-(4). . . . . \$ 100.00  ENTER APPROPRIATE BASIC FEE AMOUNT =				CALCULATIONS	PTO USE ONLY
Surcharge of \$130.00 for furnishing the oath or declaration later than 20 30 months from the earliest claimed priority date (37 CFR 1.492(e)).				\$ 0.00	
Claims	Number Filed	Number Extra	RATE		
Total Claims	25 - 20 =	5	X \$18.00	\$ 90.00	
Independent Claims	3 - 3 =	0	X \$84.00	\$ 0.00	
Multiple dependent claim(s) (if applicable)			+ \$280.00	\$ 0.00	
TOTAL OF ABOVE CALCULATIONS =				\$980.00	
Applicant claims small entity status. See 37 CFR 1.27. The fees indicated above are reduced by 1/2.				\$ 0.00	
SUBTOTAL =				\$980.00	
Processing fee of \$130.00 for furnishing the English translation later than 20 30 months from the earliest claimed priority date (37 CFR 1.492(f)).				\$ 0.00	
Extension of Time fee in the amount of \$				\$ 0.00	
TOTAL NATIONAL FEE =				\$980.00	
Fee for recording the enclosed assignment (37 CFR 1.21(h). The assignment must be accompanied by an appropriate cover sheet (37 CFR 3.28, 3.31). \$40.00 per property				\$ 0.00	
TOTAL FEES ENCLOSED =				\$980.00	
				Amount to be refunded	\$
				Charged	\$
a. <input checked="" type="checkbox"/> A check in the amount of \$ 980.00 to cover the above fees is enclosed. b. <input type="checkbox"/> Please charge my Deposit Account No. _____ in the amount of \$ _____ to cover the above fees. c. <input checked="" type="checkbox"/> The Commissioner is hereby authorized to charge any additional fees which may be required, or credit any overpayment to Deposit Account No. 19-0089  NOTE: Where an appropriate time limit under 37 CFR 1.494 or 1.495 has not been met, a petition to revive (37 CFR 1.137(a) or (b)) must be filed and granted to restore the application to pending status.  SEND ALL CORRESPONDENCE TO CUSTOMER NO. 7055 AT THE PRESENT ADDRESS OF:  Bruce H. Bernstein GREENBLUM & BERNSTEIN, P.L.C. 1941 Roland Clarke Place Reston, VA 20191 (703) 716-1191					
				Signature: <i>Bruce H. Bernstein</i> NAME: Bruce H. Bernstein 29,027 REGISTRATION NUMBER	

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## DESCRIPTION

## METHOD OF TREATING EPOXY RESIN-CURED PRODUCT

## Technical Field

5       The present invention relates to a method of treating an epoxy resin-cured product, a method of separating a composite material of inorganic matter and epoxy resin-cured product, and a treatment liquid for an epoxy resin-cured product.

10

## Background Art

      Epoxy resin-cured products are used in a variety of fields such as insulating materials, adhesives, and paints due to their superiority in electrical characteristics, heat  
15   resistance, and adhesion. However, since the epoxy resin becomes infusible and insoluble in general-purpose solvents after it is thermo-cured, it has been difficult to recycle the epoxy resin-cured products and products to which the epoxy resin-cured product adheres or on which the epoxy  
20   resin-cured product is applied. Furthermore, it has been difficult as well to separate inorganic fillers, inorganic fibers, inorganic textile fabrics, inorganic fiber nonwoven fabrics, etc., which are blended with the epoxy resin to increase the resin's mechanical properties or the like, from  
25   the resin-cured products, for recycling purpose.

      A technique of grinding, pulverizing, and separating according to specific gravity is a known example of a method

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of separating a printed wiring board, which is a composite material of epoxy resin-cured product and inorganic matter such as glass fiber and metal, into each component. However, although recovery and recycling of metals are possible to some extent in this method, pulverized glass fibers and resin powders thus obtained can be utilized only as fillers. Therefore their value as valuable goods is significantly low, and, in addition, it has been impossible to separate each component perfectly. Also, known examples include a method in which resins of printed wiring boards, laminates or the like are thermal-decomposed to recover metals and glass fibers while the thermal-decomposed resins are gasified or liquidized for the recovery, and a method of recovering not only inorganic matter but also the thermal decomposed resins. However, these methods have several problems as follows. Thermal decomposition of resins requires high temperatures so that obtained metals and inorganic matter are oxidized and denaturalized while resins are oxidized or carbonized, thereby decreasing their value; when resins contain toxic materials such as halogens and lead, separation and treatment of these materials require enormous costs.

As mentioned above, for the purpose of recycling of resins, thermal decomposition of resins is not generally preferred. For this reason, there has been a proposition for a method of dissolving thermosetting resin-cured wastes in solvents, especially in organic solvents to separate inorganic matter such as metal and glass (see Japanese



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a treatment liquid for an epoxy resin-cured product for use in the above-mentioned two methods.

In the steps of processing printed wiring boards, a variety of etching liquids have been used in order to dissolve the epoxy resin-cured product and, as an etching liquid for the epoxy resin-cured product comprising a halogenated high-molecular-weight epoxy polymer, we have reported a series of etching liquids which contain amide-based solvents and alkali metal compounds but do not contain dangerous chemicals such as concentrated sulfuric acid and chromic acid (see Japanese Unexamined Patent Application Publication No. 8-325436, No. 8-325437, No. 8-325438, No. 9-316445, and No. 10-126052). All of these etching liquids have conventionally been directed to etching and removing parts of resin-cured products to form electric circuits, and recovery and recycling of the resin component after removal have been neither meant nor intended at all. However, we have found a fact that the resin components and inorganic matter can easily be recovered in a recyclable state by using these etching liquids improved as a liquid for decomposition and dissolution of the epoxy resin-cured product.

According to one aspect of the present invention there is provided a method of treating an epoxy resin-cured product with a treatment liquid containing a decomposition catalyst for epoxy resin-cured products and an organic solvent to decompose and dissolve the epoxy resin-cured

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product. The decomposition catalyst for epoxy resin-cured products serves as a catalyst to cleave the ether linkage in the epoxy resin, thereby accelerating the decomposition of the epoxy resin swollen with the organic solvent.

5 According to another aspect of the present invention there is provided a method of separating a composite material of inorganic matter and epoxy resin-cured product, which comprises the following steps (1) and (2):

(1) treating the composite material of inorganic matter  
10 and epoxy resin-cured product with a treatment liquid containing a decomposition catalyst for epoxy resin-cured products and an organic solvent to decompose and dissolve the epoxy resin-cured product;

(2) separating the inorganic matter from the liquid  
15 obtained in the step (1).

According to another aspect of the present invention there is provided a treatment liquid for an epoxy resin-cured product containing a decomposition catalyst for epoxy resin-cured products and an organic solvent.

20 The treatment method and separation method according to the present invention can decompose and dissolve the epoxy resin-cured product to easily recover the resin components in a recyclable state without causing thermal decomposition or without requiring pulverization of materials to be  
25 treated by using the above-mentioned treatment liquid. The decomposition products can be recycled, for example, as a raw material for synthetic resins.

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When a composite material containing inorganic matter is a subject to be treated, the inorganic matter insoluble in the treatment liquid (inorganic filler, inorganic fiber, inorganic foil, etc.) can easily be separated from the treatment liquid containing the resin components, which enables recycling of the inorganic matter as well as the resin components.

Furthermore, by applying the present invention to printed wired boards with parts mounted thereon, not only inorganic fillers and inorganic fibers, but also metal foils, which are circuits, and parts connected to the circuits are dispersed in the treatment liquid by decomposition and dissolution of the resin-cured product, thereby enabling separation and recovery of such inorganic matter.

#### Best Mode for Carrying Out the Invention

A method of treating an epoxy resin-cured product according to the present invention involves treating the epoxy resin-cured product with a treatment liquid containing a decomposition catalyst for epoxy resin-cured products and an organic solvent to decompose and dissolve the epoxy resin-cured product.

The epoxy resin-cured product to be treated is essentially comprised of an epoxy resin, a curing agent, and a crosslinking agent, and may further contain a curing accelerator, a catalyst, an elastomer, a flame retardant, etc. Components other than the epoxy resin may be present



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as impurities.

Any epoxy resin can be an object to be treated as far as it has an epoxy group in the molecule, and examples thereof include bisphenol A-type epoxy resin, bisphenol F-type epoxy resin, bisphenol S-type epoxy resin, alicyclic epoxy resin, aliphatic linear epoxy resin, phenol novolak-type epoxy resin, cresol novolak-type epoxy resin, bisphenol A novolak-type epoxy resin, diglycidyl ethers of biphenols, diglycidyl ethers of naphthalenediols, diglycidyl ethers of phenols, diglycidyl ethers of alcohols, and compounds  
10 obtained from the above-mentioned compounds by substitution with an alkyl group, halogenation, or hydrogenation. They can be used singly or in combination.

Among the above-mentioned epoxy resins, use can  
15 preferably be made of epoxy resins whose hydrogen located in the ortho position on the benzene ring with respect to the ether group bonded to the same benzene group is substituted with a halogen atom such as chlorine and bromine. Such epoxy resins are obtained by reacting halogenated bisphenol  
20 compounds such as tetrabromobisphenol A with epichlorohydrin, and examples include halogenated bisphenol A-type epoxy resin, halogenated bisphenol F-type epoxy resin, and halogenated bisphenol S-type epoxy resin. When an electron-withdrawing halogen atom is bonded to the benzene ring of  
25 the epoxy resin, cleavage of the ether linkage is more likely to occur, which can facilitate decomposition and dissolution of the epoxy resin-cured product with a

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treatment liquid.

Any curing agent for epoxy resins can be included as far as it can cure epoxy resins, and examples thereof include multifunctional phenols, amines, imidazole compounds, acid anhydrides, organic phosphorus compounds, and halides of the above-mentioned compounds. Any of these compounds can be used singly or in combination. Although the amount of the curing agent to be blended is not specifically limited as far as it can promote the curing reaction of the epoxy group, the curing agent is preferably included in a range of 0.01 to 5.0 equivalents, more preferably in a range of 0.8 to 1.2 equivalents, relative to one mole of the epoxy group.

Examples of the multifunctional phenols include monocyclic bifunctional phenols such as hydroquinone, resorcinol, and catechol; polycyclic bifunctional phenols such as bisphenol A, bisphenol F, naphthalenediols, and biphenols; and these compounds halogenated or substituted with an alkyl group. Novolaks and resols, which are polycondensates of these phenols and aldehydes, can be used as well.

Examples of the amines include aliphatic or aromatic primary amines, aliphatic or aromatic secondary amines, aliphatic or aromatic tertiary amines, quaternary ammonium salts thereof, alicyclic amines, guanidines, and urea derivatives. They can be, for example, any one of N,N-benzyl dimethylamine, 2-(dimethylaminomethyl)phenol, 2,4,6-

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tris(dimethylaminomethyl)phenol, tetramethylguanidine,  
triethanolamine, N,N'-dimethylpiperazine, 1,4-  
diazabicyclo[2.2.2]octane, 1,8-diazabicyclo[5.4.0]-7-  
undecene, 1,5-diazabicyclo[4.4.0]-5-nonene,  
5 hexamethylenetetramine, pyridine, picoline, piperidine,  
pyrrolidine, dimethylcyclohexylamine, dimethylhexylamine,  
cyclohexylamine, diisobutylamine, di-n-butylamine,  
diphenylamine, N-methylaniline, tri-n-propylamine, tri-n-  
octylamine, tri-n-butylamine, triphenylamine,  
10 tetramethylammonium chloride, tetramethylammonium bromide,  
tetramethylammonium iodide, triethylenetetramine,  
diaminodiphenylmethane, diaminodiphenyl ether, dicyandiamide,  
tolylbiguanide, guanylurea, and dimethylurea.

Examples of the imidazole compounds include imidazole,  
15 2-ethylimidazole, 2-ethyl-4-methylimidazole, 2-  
methylimidazole, 2-phenylimidazole, 2-undecylimidazole, 1-  
benzyl-2-methylimidazole, 2-heptadecylimidazole, 4,5-  
diphenylimidazole, 2-methylimidazoline, 2-phenylimidazoline,  
2-undecylimidazoline, 2-heptadecylimidazoline, 2-  
20 isopropylimidazole, 2,4-dimethylimidazole, 2-phenyl-4-  
methylimidazole, 2-ethylidazoline, 2-phenyl-4-  
methylimidazoline, benzimidazole, and 1-cyanoethylimidazole.

Examples of the acid anhydrides include phthalic  
anhydride, hexahydrophthalic anhydride, pyromellitic  
25 dianhydride, and benzophenonetetracarboxylic dianhydride.

Any organic phosphorus compounds can be used as far as  
it is a phosphorus compound having an organic group.

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Examples thereof include hexamethylphosphoric triamide, tri(dichloropropyl) phosphate, tri(chloropropyl) phosphate, triphenyl phosphite, trimethyl phosphate, phenylphosphonic acid, triphenylphosphine, tri-n-butylphosphine, and  
5 diphenylphosphine.

The epoxy resin-cured product may contain a curing accelerator. Typical examples of curing accelerators include, but are not limited to, tertiary amines, imidazoles, and quaternary ammonium salts.

10 The epoxy resin-cured product to be subjected to the treatment according to the present invention can be obtained by curing the epoxy resin composition comprising the above-described components using an arbitrary known method and any curing condition can be selected as far as the reaction  
15 proceeds. For example, any temperature may be used as far as the reaction proceeds, although curing is in general preferably performed at a temperature ranging between room temperature and about 250°C. Furthermore, the curing reaction may be carried out either under pressure, under  
20 atmospheric pressure, or under reduced pressure.

Next, the treatment liquid to decompose and dissolve the above-mentioned resin-cured products contains a decomposition catalyst for epoxy resin-cured products and an organic solvent. In other words, the treatment liquid used  
25 in the present invention contains, as essential components, at least a compound that serves as a catalyst for cleavage of the ether linkage of the epoxy resin-cured products and

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an organic solvent. The treatment liquid may further contain arbitrary known compounds other than the two components or even impurities.

In a preferred embodiment, the decomposition catalyst  
5 for epoxy resin-cured products contains one or more compounds selected from alkali metals and/or alkali metal compounds, phosphorous-containing acids and/or salts thereof, and organic acids and/or salts thereof. The term "one or more compounds" means any combination of these compounds is  
10 acceptable and examples thereof include a combination of alkali metal compounds, a combination of phosphorous-containing acids (salts), a combination of an alkali metal compound and a phosphorous-containing acid (salt), and a combination of a phosphorous-containing acid (salt) and an  
15 organic acid (salt). These compounds are preferably added in an amount of 0.001 to 80 % by weight, especially 0.1 to 30 % by weight in total in the organic solvent. When the amount is less than 0.001 % by weight, the decomposition rate of the epoxy resin-cured product tends to be lower,  
20 while when the amount exceeds 80 % by weight, the preparation of the treatment liquid tends to be difficult. Moreover, it is not always necessary that all of these compounds are dissolved in the organic solvent, and a saturated solution in which the solute is in an equilibrium  
25 state in the solution is also useful because insoluble parts in the saturated solution can supply a source of fresh compounds when the already-dissolved compounds become

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deactivated during the reaction.

Examples of the alkali metals include lithium, sodium, potassium, rubidium, and cesium. Examples of the alkali metal compounds include hydrides, hydroxides, borohydrides, amide compounds, fluorides, chlorides, bromides, iodides, borates, phosphates, carbonates, sulfates, nitrates, organic acid salts, alcoholates, and phenolates, obtained from any one of these alkali metals. Among these compounds, the alkali metal salts are preferably used due to their high solubility in organic solvents, high catalytic effect (ion activity), and low toxicity as ions. Any of these metals and metal compounds can be used singly or in combination and preferably added in an amount of 0.01 to 80 % by weight, especially 0.1 to 10 % by weight in an organic solvent.

Examples of the phosphorous-containing acids used herein include phosphoric acid, metaphosphoric acid, hypophosphoric acid, phosphorous acid (phosphonic acid), hypophosphorous acid (phosphinic acid), pyrophosphoric acid, trimetaphosphoric acid, tetrametaphosphoric acid, and pyrophosphorous acid. The salts of phosphorous-containing acids used herein are salts of anions of the above-mentioned phosphorous-containing acids and cations, and the cations can be, for example, ions of lithium, sodium, potassium, rubidium, cesium, beryllium, magnesium, calcium, strontium, barium, titanium, zirconium, vanadium, chromium, manganese, iron, cobalt, nickel, copper, silver, palladium, zinc, aluminum, gallium, stannum, or ammonium. Among these

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compounds, alkali metal compounds are preferred for the same reason as described above and hydrates are preferred in terms of solubility in solvents. These salts can be any one of primary salts having a metal and two hydrogens, secondary salts having two metals and a hydrogen, and a tertiary salts having three metals. Also, these salts can be any one of acidic salts, alkaline salts, and neutral salts. Any of these compounds can be used singly or in combination and preferably added in an amount of 0.01 to 80 % by weight, especially 0.1 to 30 % by weight in an organic solvent.

Examples of the organic acids include acrylic acid, adipic acid, ascorbic acid, asparagic acid, aminobenzoic acid, alginic acid, benzoic acid, oleic acid, formic acid, citric acid, glycolic acid, gluconic acid, glutamic acid, cinnamic acid, succinic acid, acetic acid, salicylic acid, oxalic acid, tartaric acid, toluenesulfonic acid, nicotinic acid, lactic acid, uric acid, halogenated acetic acid, phthalic acid, benzenesulfonic acid, malonic acid, butyric acid, and malic acid. The salts of organic acids are salts of anions of the above-mentioned organic acids and cations other than hydrogen, and the cations can be, for example, ions of lithium, sodium, potassium, rubidium, cesium, beryllium, magnesium, calcium, strontium, barium, titanium, zirconium, vanadium, chromium, manganese, iron, cobalt, nickel, copper, silver, palladium, zinc, aluminum, gallium, stannum, or ammonium. Among these compounds, hydrates and alkali metal salts are preferred for the same reason as

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described above. These salts can be any one of primary salts having a metal and two hydrogens, secondary salts having two metals and a hydrogen, and a tertiary salts having three metals. Also, these salts can be any one of  
5 acidic salts, alkaline salts, and neutral salts. Any of these compounds can be used singly or in combination and preferably added in an amount of 0.01 to 80 % by weight, especially 0.1 to 30 % by weight in an organic solvent.

As an organic solvent, one or more solvents selected  
10 from amide-based, alcohol-based, ketone-based, and ether-based solvents are preferably used in terms of solubility of ionic catalysts, but use can also be made of other solvents such as hydrocarbons, halogenated hydrocarbons, phenols, acetals, fatty acids, acid anhydrides, esters, nitrogen  
15 compounds, sulfur compounds (such as dimethylsulfoxide) and solvents having two or more functional groups (ester and ether, alcohol and ether, and the like). Any of these solvents can be used singly or in combination (for example, a combination of amide-based solvents and a combination of  
20 an amide-based solvent and a non-amide-based solvent). Furthermore, any solvent can be concurrently employed and inorganic solvents such as water and ammonia may be mixed, and impurities may be present.

Preferred examples of the amide-based solvents include  
25 formamide, N-methylformamide, N,N-dimethylformamide, N,N-diethylformamide, acetamide, N-methylacetamide, N,N-dimethylacetamide, N,N,N',N'-tetramethylurea, 2-pyrrolidone,



N-methyl-2-pyrrolidone, caprolactam, and carbamates.

Examples of the alcohol-based solvents include methanol, ethanol, 1-propanol, 2-propanol, 1-butanol, 2-butanol, isobutanol, tert-butanol, 1-pentanol, 2-pentanol, 3-pentanol, 2-methyl-1-butanol, iso-pentyl alcohol, tert-pentyl alcohol, 3-methyl-2-butanol, neopentyl alcohol, 1-hexanol, 2-methyl-1-pentanol, 4-methyl-2-pentanol, 2-ethyl-1-butanol, 1-heptanol, 2-heptanol, 3-heptanol, cyclohexanol, 1-methylcyclohexanol, 2-methylcyclohexanol, 3-methylcyclohexanol, 4-methylcyclohexanol, ethylene glycol, ethylene glycol monomethyl ether, ethylene glycol monoethyl ether, ethylene glycol monopropyl ether, ethylene glycol monobutyl ether, diethylene glycol, diethylene glycol monomethyl ether, diethylene glycol monoethyl ether, diethylene glycol monopropyl ether, diethylene glycol monobutyl ether, triethylene glycol, triethylene glycol monomethyl ether, triethylene glycol monoethyl ether, tetraethylene glycol, polyethylene glycol (molecular weight 200 to 400), 1,2-propanediol, 1,3-propanediol, 1,2-butanediol, 1,3-butanediol, 1,4-butanediol, 2,3-butanediol, 1,5-pentanediol, glycerol, and dipropylene glycol.

Examples of the ketone-based solvents include acetone, methyl ethyl ketone, 2-pentanone, 3-pentanone, 2-hexanone, methyl isobutyl ketone, 2-heptanone, 4-heptanone, diisobutyl ketone, cyclohexanone, methylcyclohexanone, phorone, and isophorone.

Examples of the ether-based solvents include dipropyl

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ether, diisopropyl ether, dibutyl ether, dihexyl ether, anisole, phenetol, dioxane, tetrahydrofuran, acetal, ethylene glycol dimethyl ether, ethylene glycol diethyl ether, diethylene glycol dimethyl ether, and diethylene glycol diethyl ether.

Although any temperature can be used in preparing the treatment liquid, a temperature ranging between the melting point and the boiling point of the solvent used is preferred. Moreover, the treatment liquid may be prepared either in an air atmosphere or in an inert gas and either under atmospheric pressure (normal pressure), under reduced pressure, or under pressure. Other components such as surfactant may be added to the treatment liquid thus obtained.

The treatment method involving decomposition and dissolution of the epoxy resin-cured product, which is a subject to be treated, with the above-mentioned treatment liquid is not specifically limited, and the treatment may be performed, for example, by immersing the epoxy resin-cured product in the treatment liquid or by spraying the treatment liquid on the epoxy resin-cured product instead of immersing. The treatment liquid may be stirred by an agitator, a pump, or blowing of gas. In the case of dipping treatment, the treatment may be conducted while vibrating the treatment liquid by ultrasound. The treatment liquid for the treatment of the resin-cured product may be used at an arbitrary temperature ranging between the freezing point and

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the boiling point of the solvent in order to control the treatment rate. For the purpose of avoiding undesired thermal decomposition of the cured resin, however, the treatment liquid may preferably have a temperature of 250°C or lower in air and 300°C or lower in an inert gas. The treatment liquid may be used and stored either in air or in an inert gas, and either under atmospheric pressure, under reduced pressure, or under pressure. Although high pressure may be effective in some cases, atmospheric pressure is more preferred when safety is of concern.

The size of the subject to be treated is not specifically limited, and the subject may be just in the state as it is disposed of and recovered (for example, a 250 mm-square printed wiring board) or may be crushed and pulverized. A crushed and pulverized subject requires a shorter time to be treated, but the recovered materials thus obtained have limited applications. In view of this, the size of crushed pieces is preferably approximately 5 mm or larger, or 10 mm or larger. Moreover, when the crushing degree of the subject to be treated reaches pulverization (approximately 1 mm or smaller), recovered glass fibers, for example, find no application for recycling, which may significantly decrease their value. The amount of the treatment liquid with respect to the subject is not specifically limited, and if the treatment liquid contacts the subject, such an amount is enough.

Methods of separating and washing, and application of

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the resin components (the organic components derived from the resin) decomposed and dissolved in this manner are not specifically limited. For example, the resin components can be obtained by separating and removing residues (insoluble parts) from the liquid by precipitation after the treatment and removing the organic solvent by distillation. The components recovered can be recycled as a raw material for synthetic resins.

Specifically, the decomposition products can be recycled in their recovered state or preferably after purification, when the decomposition products obtained after the treatment of the resin-cured product are, for example, phenols, glycidyl ethers of phenols, metal salts of phenols, amines, carboxylic acids, and halogenated or hydrogenated products of the above-mentioned compounds (in the concrete, such as phenol, cresol, dimethylphenol, propylphenol, ethylphenol, hydroquinone, resorcinol, catechol, bisphenol A, bisphenol F, biphenol, dihydroxydiphenyl ether, dihydroxydiphenyl sulfone, phenol novolak, cresol novolak, bisphenol A novolak, and glycidyl ethers of these compounds, halides of these compounds, alkali metal salts of these compounds, and ammonium salts of these compounds).

Next, a method of separating a composite material of inorganic matter and epoxy resin-cured product according to the present invention includes the following steps (1) and (2);

(1) treating the composite material of inorganic matter

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and epoxy resin-cured product with a treatment liquid containing a decomposition catalyst for epoxy resin-cured products and an organic solvent to decompose and dissolve the epoxy resin-cured product; and

- 5       (2) separating the inorganic matter from the liquid obtained in the step (1).

Composite materials of inorganic matter and epoxy resin-cured product to be treated comprise the above-mentioned epoxy resin-cured product and inorganic matter,  
10 and are typically insulating boards, metal-clad laminates, or printed wiring boards, which contain metals and glass as inorganic matter.

The inorganic matter can be, for example, metals and metal oxides, hydroxides, halides, or nitrides. More  
15 concretely, examples include boron, aluminum, iron, silicon, titanium, chrome, cobalt, nickel, zinc, palladium, silver, stannum, tungsten, platinum, gold, lead, alumina, zirconia, titania, magnesia, silicon carbide, silicon nitride, boron nitride, mica, silica, clay, glass, carbon, calcium  
20 carbonate, aluminum hydroxide, magnesium hydroxide, and calcium silicate. These metals and compounds may be fused or mixed. The shape of the inorganic matter may be, for example, powder, granule, fiber, foil, film, wire, or circuit. The fiber may be in a mat-like state or woven like  
25 fabric. Although the content of inorganic matter in a composite material is arbitrary, it generally ranges from 5 to 90 % by weight based on the total amount of the composite

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material.

When the inorganic matter is, for example, powdery, granular, or staple-like, such inorganic matter is blended with the above-mentioned epoxy resin composition for curing, which is cured by cast molding or the like to obtain a composite material of inorganic matter and epoxy resin-cured product. When the inorganic matter is mat-like or fabric-like, such inorganic matter is immersed in the above-mentioned epoxy resin composition (which may contain powdery, granular, or staple-like inorganic matter as described above) and then partial-cured to obtain a B-stage prepreg. Such a B-stage prepreg and laminates obtained by laminating and curing this prepreg or a prepreg at a pre-B-stage are also the composite material of inorganic matter and epoxy resin-cured product. The present invention is further applicable to the composite materials of inorganic matter and epoxy resin-cured product such as: resin boards or films covered with a metal foil obtained by coating or flow-coating the metal foil with the above-mentioned epoxy resin composition (which may contain powdery, granular, or staple-like inorganic matter as described above) and then curing the same; resin boards covered with a metal foil obtained by laminating a metal foil onto the above-mentioned prepreg or prepreg at a pre-B-stage and then curing the same; laminates covered with a metal foil obtained by laminating the above-mentioned metal foil-covered resin boards or films (before curing) or metal foil-covered resin boards (which are formed

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with prepregs and before curing) onto the above-mentioned laminates and then curing the same; interlayer circuit boards obtained by using the above-mentioned epoxy resin composition (which may contain powdery, granular, or staple-  
5 like inorganic matter as described above) and metal foils and applying an appropriate circuit formation method; and laminates covered with a metal foil obtained by laminating the above-mentioned metal foil covered-resin boards or films (before curing) or the metal foil covered-resin boards  
10 (which are formed with prepregs and before curing) onto the above-mentioned interlayer circuit boards and then curing the same. The above-mentioned resin boards, films, or laminates, which have metal foils, may be ones having a circuit formed by applying a circuit formation method to  
15 metal foils.

The step (1) of treating the composite materials, the resin components thus obtained and recycling thereof are the same as described for the treatment method of the epoxy resin-cured product. The separation step (2) is not  
20 specifically limited and inorganic matter such as metals and glass can easily be separated from the liquid, for example, by filtration or decantation, for recycling purpose. According to the separation method of the present invention, because subjects to be treated such as substrate need not to  
25 be crushed or pulverized before the treatment, inorganic matter can be recovered in such a recyclable state that is suitable for a variety of applications. For example, it is







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are as follows: NMP, N-methyl-2-pyrrolidone; CHON, cyclohexanone; PEG, polyethylene glycol #200; DGMM, diethylene glycol monomethyl ether; and DGDM, diethylene glycol dimethyl ether.

5 Treatment liquid that contained no alkali metal compound was used in each case of Comparative Examples A1 to A5, and a 3% aqueous solution of sodium hydroxide was prepared in Comparative Example A6.

(Treatment of epoxy resin-cured product)

10 The two resin boards were cut into pieces of a size of about 5 mm square each and 0.5 g of the pieces were weighed out and placed in 50 g of each treatment liquid. The treatment liquids were maintained at a predetermined temperature (60°C, 80°C, or 100°C) and agitated hard for one  
15 hour. The treatment liquids were visually inspected to evaluate the solubility according to the following rating.  
5: all dissolved; 4: almost all dissolved; 3: half dissolved; 2: significantly broken; 1: broken to some extent; 0: no change.

20 The results are shown in Table 1.

[Table 1] A. Solubility in treatment liquid containing alkali metal compound (1)

Example	Composition of treatment liquid	Bromine content : 20 % by weight			Bromine content : 30 % by weight		
		60 °C	80 °C	100 °C	60 °C	80 °C	100 °C
A 1	KOH	NMP	4	5	5	5	5
A 2		CHON	2	3	4	4	5
A 3		PEG	0	1	2	3	5
A 4		DGMM	1	1	2	3	5
A 5		DGDM	0	1	3	4	5
A 6	NaCl	NMP	3	4	5	4	5
A 7		CHON	1	2	3	2	5
A 8		PEG	0	0	1	1	2
A 9		DGMM	0	1	2	2	3
A 10		DGDM	0	1	3	2	3
A 11	Na2CO3	NMP	1	3	5	3	5
A 12		CHON	0	2	2	2	4
A 13		PEG	0	0	1	1	2
A 14		DGMM	0	1	2	2	3
A 15		DGDM	0	1	2	2	3
Comp. Ex.							
A 1	—	NMP	0	0	0	0	0
A 2		CHON	0	0	0	0	0
A 3		PEG	0	0	0	0	0
A 4		DGMM	0	0	0	0	0
A 5		DGDM	0	0	0	0	0
A 6	KOH	Water	0	0	0	0	0

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As shown in Table 1, when the resin boards were treated with an organic solvent alone as in Comparative Examples A1 to A5, the resin-cured products were not dissolved. In Comparative Example A6, where the resin boards were treated with an aqueous solution of an alkali metal compound, the result was the same.

In contrast, in Examples A1 to A15, when the resin boards were treated with an organic solvent solution of an alkali metal compound, solubility was good in all the cases and the resin was dissolved completely in some cases.

B. Solubility in treatment liquid containing alkali metal compound (2)

[EXAMPLES B1 to B15, COMPARATIVE EXAMPLES B1 to B6]

(Preparation of composite material of inorganic matter and epoxy resin-cured product)

(1) The following two kinds of copper-clad laminate were provided: "MCL-E-679" (bromine content in the resin-cured product is 15%) and "MCL-E-67" (bromine content in the resin-cured product is 20%); both which are manufactured from a brominated epoxy resin, a bromine-free epoxy resin, a curing agent, a curing accelerator, glass cloth, and copper foil, and both are product names of Hitachi Chemical Co., Ltd.

(2) An epoxy resin composition was prepared by blending a brominated bisphenol A-type epoxy resin (manufactured by Tohto Kasei Co., Ltd., product name "AER8011"; epoxy

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equivalent 470, bromine content 20%), a high-molecular-weight brominated bisphenol A-type epoxy resin (bromine content 53%, number average molecular weight 25,000; the number average molecular weight was measured by gel permeation chromatography and converted based on a calibration curve prepared with a standard polystyrene), a phenol novolak resin (manufactured by Hitachi Chemical Co., Ltd., product name "HP850N"; hydroxyl equivalent 106) as a curing agent, and 2-methyl-4-methylimidazole as a curing accelerator. The amounts of the two epoxy resins and the phenol novolak resin were adjusted such that the ratio of the epoxy equivalent to the hydroxyl equivalent was 1. The amount of each epoxy resin was adjusted to give a bromine content of 28% relative to the total amount of the epoxy resin composition excluding the amount of inorganic matter. The curing agent was added in an amount of 0.5% based on the total amount of the epoxy resins.

Glass cloth was impregnated with the epoxy resin composition obtained and predried at 160°C for 4 minutes to form a prepreg. A copper foil was laminated onto the prepreg, which was then dried at 170°C for 90 minutes to obtain a copper-clad laminate, which is a composite material of inorganic matter and epoxy resin-cured product. This is referred to as "self-made laminate sample" hereinafter.

(3) Two kinds of the copper-clad laminate of (1) and the self-made laminate sample of (2) were cut into pieces of a size of 10 mm by 50 mm each and three lines of circuit of

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1 mm-wide copper foils were formed on the both sides of each piece to obtain test pieces.

(Preparation of treatment liquid)

In each case of Examples B1 to B15, components listed  
5 in Table 2 were mixed to prepare a 3% solution of an alkali metal compound in an amide-based, ketone-based, alcohol-based or ether-based organic solvent(s). The combinations shown in Examples B1, B2, and B5 were NMP 80% and PEG 17%, CHON 80% and DGMM 17%, and DGDM 80% and PEG 17%,  
10 respectively. In some of these treatment liquids, the alkali metal compound was not completely dissolved and precipitated on the bottom of a tank when the liquids were allowed to stand. The compound names in Table 2 are the same as in Table 1.

15 Treatment liquid that contained no alkali metal compound was used in each case of Comparative Examples B1 to B5, and a 3% aqueous solution of sodium hydroxide was prepared in Comparative Example B6.

(Treatment of composite material)

20 The above-mentioned three kinds of test pieces were weighed and then immersed in each of the treatment liquids at a predetermined temperature (60°C or 100°C). After 60 minutes, the test pieces were taken out from the treatment liquid and weighed again. The weight of the resin-cured  
25 product before the treatment was calculated from the ratio of the resin in the composite material, and the rate of change in weight of the resin-cured product was calculated



[Table 2] B. Solubility in treatment liquid containing alkali metal compound (2)  
(Solubility /%)

Example	Composition of treatment liquid	Temperature : 60 °C				Temperature : 100 °C			
		E-679 Br 15wt%	E-67 Br 20wt%	Sample* Br 28wt%	E-679 Br 15wt%	E-67 Br 20wt%	Sample* Br 28wt%	E-679 Br 15wt%	Sample* Br 28wt%
B 1	NMP, PEG	1.3	4.9	48.0	2.5	29.7	93.5		
B 2	CHON, DGMM	2.1	2.2	12.3	2.9	5.4	57.8		
B 3	PEG	2.2	2.3	3.3	2.3	6.1	33.7		
B 4	DGMM	1.7	2.1	9.6	2.6	16.3	89.0		
B 5	DGDM, PEG	1.8	2.1	12.7	2.9	5.4	57.8		
B 6	NMP	0.9	3.5	25.1	3.2	4.5	68.0		
B 7	CHON	1.9	1.7	11.6	2.0	2.4	23.8		
B 8	PEG	2.1	1.8	3.4	2.7	2.1	4.2		
B 9	DGMM	1.7	1.2	7.2	2.3	1.8	8.3		
B 10	DGDM	1.5	1.8	9.4	1.6	2.8	9.8		
B 11	NMP	0.9	1.8	25.4	1.9	23.4	53.3		
B 12	CHON	1.8	1.7	9.7	1.9	3.7	22.6		
B 13	PEG	2.1	1.5	3.3	1.7	1.9	3.1		
B 14	DGMM	2.0	2.7	9.6	2.3	3.2	10.0		
B 15	DGDM	2.2	1.6	13.4	2.0	3.0	13.8		
Comp. Ex.									
B 1	NMP	0.4	0.2	0.6	0.8	0.5	0.9		
B 2	CHON	0.3	0.3	0.4	—	—	—		
B 3	PEG	0.2	0.2	0.3	—	—	—		
B 4	DGMM	0.4	0.3	0.5	—	—	—		
B 5	DGDM	0.5	0.4	0.4	—	—	—		
B 6	Water	0.3	0.1	0.2	0.2	0.4	0.8		

\* self-made laminate sample





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(Preparation of treatment liquid)

In each case of Examples C1 to C45, a phosphorous-containing acid or its salt listed in Table 3 was weighed in an amount corresponding to 1.0 equivalent with respect to one liter of solvent and mixed with the solvent at room temperature to prepare a treatment liquid. The compound names of the organic solvents shown in Table 3 are the same as in Examples A.

Comparative Examples are: Comparative Examples C1 to C3 where each treatment liquid contained an organic solvent only; Comparative Examples C4 where the treatment liquid contained water only; and Comparative Examples C5 to C21 where each treatment liquid was a 3% aqueous solution of a phosphorous-containing acid or its salt listed in Table 3.

(Preparation of composite material)

Each of above-mentioned treatment liquids was placed in a flask equipped with a condenser, a thermometer, an inlet for nitrogen, and a stirrer and heated to 140°C (for organic solvents) or 100°C (for aqueous solutions) in an oil bath while being mildly stirred under nitrogen gas stream. The above-mentioned test piece was weighed, immersed in each treatment liquid for four hours, taken out and weighed again. The change in weight before and after the treatment was divided by the total weight of the resin in the test piece which was weighted beforehand to obtain a solubility of the resin-cured product.

The results are shown in Table 3.

[illegible]

Ex.	Composition	%	Ex.	Composition	%	Comp. Ex.	Composition	%		
C 1	HPO <sub>3</sub>	NMP	3.6	C 24	Na <sub>2</sub> HPO <sub>3</sub>	NMP	6.5	C 1	NMP	0.8
C 2		DGMM	1.2	C 25	·5H <sub>2</sub> O	DGMM	2.2	C 2	DGMM	1.2
C 3		CHON	0.4	C 26		CHON	0.8	C 3	CHON	0.9
C 4	H <sub>3</sub> PO <sub>4</sub>	NMP	13.9	C 27	NaPH <sub>2</sub> O <sub>2</sub>	NMP	11.0	C 4	W (=Water)	0.0
C 5		DGMM	0.3	C 28	·H <sub>2</sub> O	DGMM	1.6	C 5	HPO <sub>3</sub> + W	0.0
C 6		CHON	0.1	C 29		CHON	0.9	C 6	H <sub>3</sub> PO <sub>4</sub> + W	0.0
C 7	H <sub>3</sub> PO <sub>3</sub>	NMP	11.4	C 30	K <sub>3</sub> PO <sub>4</sub> ·	NMP	20.5	C 7	H <sub>3</sub> PO <sub>3</sub> + W	0.0
C 8		DGMM	0.2	C 31	nH <sub>2</sub> O	DGMM	33.7	C 8	Na <sub>3</sub> PO <sub>4</sub> ·12H <sub>2</sub> O + W	0.0
C 9		CHON	0.1	C 32		CHON	6.4	C 9	(NaPO <sub>3</sub> ) <sub>n</sub> + W	0.0
C 10	Na <sub>3</sub> PO <sub>4</sub> ·12H <sub>2</sub> O	NMP	37.6	C 33	(KPO <sub>3</sub> ) <sub>n</sub>	NMP	3.8	C 10	Na <sub>2</sub> HPO <sub>4</sub> + W	0.0
C 11		DGMM	28.1	C 34		DGMM	1.1	C 11	Na <sub>2</sub> HPO <sub>4</sub> ·12H <sub>2</sub> O + W	0.0
C 12		CHON	4.5	C 35		CHON	0.2	C 12	NaH <sub>2</sub> PO <sub>4</sub> ·2H <sub>2</sub> O + W	0.0
C 13	(NaPO <sub>3</sub> ) <sub>n</sub>	NMP	2.1	C 36	Mg <sub>3</sub> PO <sub>4</sub>	NMP	7.4	C 13	Na <sub>2</sub> HPO <sub>3</sub> ·5H <sub>2</sub> O + W	0.0
C 14		DGMM	1.1	C 37	·8H <sub>2</sub> O	DGMM	1.5	C 14	NaPH <sub>2</sub> O <sub>2</sub> ·H <sub>2</sub> O + W	0.0
C 15		CHON	0.5	C 38	AlPO <sub>4</sub>	NMP	2.2	C 15	K <sub>3</sub> PO <sub>4</sub> ·nH <sub>2</sub> O + W	0.0
C 16	Na <sub>2</sub> HPO <sub>4</sub>	NMP	2.5	C 39		DGMM	1.8	C 16	(KPO <sub>3</sub> ) <sub>n</sub> + W	0.0
C 17		DGMM	2.3	C 40	Ca <sub>3</sub> (PO <sub>4</sub> ) <sub>2</sub>	NMP	1.8	C 17	Mg <sub>3</sub> PO <sub>4</sub> ·8H <sub>2</sub> O + W	0.0
C 18		CHON	0.3	C 41		DGMM	1.1	C 18	AlPO <sub>4</sub> + W	0.0
C 19	Na <sub>2</sub> HPO <sub>4</sub> ·12H <sub>2</sub> O	NMP	6.1	C 42	FePO <sub>4</sub>	NMP	3.7	C 19	Ca <sub>3</sub> (PO <sub>4</sub> ) <sub>2</sub> + W	0.0
C 20		CHON	1.0	C 43	·nH <sub>2</sub> O	DGMM	1.2	C 20	FePO <sub>4</sub> ·nH <sub>2</sub> O + W	0.0
C 21	NaH <sub>2</sub> PO <sub>4</sub> ·2H <sub>2</sub> O	NMP	13.0	C 44	(NH <sub>4</sub> ) <sub>3</sub> PO <sub>4</sub>	NMP	5.8	C 21	(NH <sub>4</sub> ) <sub>3</sub> PO <sub>4</sub> ·3H <sub>2</sub> O + W	0.0
C 22		DGMM	1.2	C 45	·3H <sub>2</sub> O	DGMM	1.6			
C 23		CHON	0.4							

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As shown in Table 3, the solubility was approximately 1% in Comparative Examples C1 to C3 where the treatment liquid was only an organic solvent. And the solubility was 0% in Comparative Examples C4 to C21 where the treatment liquid was only water or an aqueous solution of a phosphorous-containing acid or its salt.

In contrast, the treatment liquid dissolved the resin-cured product in all the cases of Examples C1 to C45 where the treatment liquid was an organic solvent solution of a phosphorous-containing acid or its salt although the solubility varied depending on the solvent. In particular, high solubility was shown in Examples C10, C11, C30, and C31.

D. Solubility in treatment liquid containing organic acid or its salt

[EXAMPLES D1 to D34, COMPARATIVE EXAMPLES D1 to D20]

A composite material of inorganic matter and epoxy resin-cured product was prepared as in Examples C and subjected to the same treatment as in Examples C with a respective one of the treatment liquids containing the components listed in Table 4 to determine the solubility in each treatment liquid.

The results are shown in Table 4.

(Solubility / %)

Ex.	Composition	%	Comp. Ex.	Composition	%
D1	Lithium acetate	NMP	D1	NMP	8.2
D2		DGMM	D2	DGMM	2.1
D3	Lithium acetate dihydrate	NMP	D3	W (=Water)	21.3
D4		DGMM	D4	Lithium acetate + W	3.2
D5	Lithium citrate	NMP	D5	Lithium acetate dihydrate + W	7.6
D6	tetrahydrate	DGMM	D6	Lithium citrate tetrahydrate + W	12.3
D7	Sodium acetate	NMP	D7	Sodium acetate + W	7.4
D8		DGMM	D8	Sodium benzoate + W	4.2
D9	Sodium benzoate	NMP	D9	Sodium citrate dehydrate + W	12.5
D10		DGMM	D10	Sodium succinate hexahydrate + W	1.3
D11	Sodium citrate dehydrate	NMP	D11	Sodium tartrate dihydrate + W	7.0
D12		DGMM	D12	Sodium benzenesulfonate hydrate+W	2.1
D13	Sodium succinate hexahydrate	NMP	D13	Potassium acetate + W	9.3
D14		DGMM	D14	Potassium benzoate + W	2.0
D15	Sodium tartrate dihydrate	NMP	D15	Potassium citrate + W	6.5
D16		DGMM	D16	Aluminium acetate hydrate + W	2.4
D17	Sodium benzenesulfonate hydrate	NMP	D17	Calcium acetate hydrate + W	1.9
D18		DGMM	D18	Iron citrate hydrate + W	1.3
D19	Potassium acetate	NMP	D19	Ammonium acetate + W	22.0
D20		DGMM	D20	Tartaric acid + W	13.3
D21	Potassium benzoate	NMP			10.2
D22		DGMM			1.6
D23	Potassium citrate	NMP			5.6
D24		DGMM			2.1
D25	Aluminium acetate hydrate	NMP			3.5
D26		DGMM			1.7
D27	Calcium acetate hydrate	NMP			7.3
D28		DGMM			1.6
D29	Iron citrate hydrate	NMP			3.3
D30		DGMM			2.1
D31	Ammonium acetate	NMP			6.4
D32		DGMM			1.4
D33	Tartaric acid	NMP			1.3
D34		DGMM			1.9

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As shown in Table 4, the solubility was approximately 1% in Comparative Examples D1 and D2 where the treatment liquid was an organic solvent. The solubility was 0% in Comparative Examples D3 to D20 where the treatment liquid  
5 was only water or an aqueous solution of an organic acid or its salt.

In contrast, the treatment liquid dissolved the resin-cured product in all the cases of Examples D1 to D34 where the treatment liquid was an organic solvent solution of an  
10 organic acid or its salt, although the solubility varied depending on the solvent. In particular, high solubility was shown in Examples D3, D6, D9, D19, D20, and D21.

The present disclosure relates to subject matter  
15 contained in Japanese Patent Applications NO. HEI11-286187, filed on October 7, 1999, and No. HEI11-286188, filed on October 7, 1999, the disclosure of which is incorporated herein by reference.

It is to be noted that, besides those already mentioned  
20 above, many modifications and variations of the above embodiments may be made without departing from the novel and advantageous features of the present invention. Accordingly, all such modifications and variations are intended to be included within the scope of the appended claims.

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# CLAIMS

1. A method of treating an epoxy resin-cured product with a treatment liquid containing a decomposition catalyst for epoxy resin-cured products and an organic solvent to decompose and dissolve the epoxy resin-cured product.
2. The method according to claim 1, wherein the decomposition catalyst for epoxy resin-cured products contains one or more compounds selected from alkali metals and/or alkali metal compounds, phosphorous-containing acids and/or salts thereof, and organic salts and/or salts thereof.
3. The method according to claim 1, wherein the organic solvent contains one or more solvents selected from amide-based solvents, alcohol-based solvents, ketone-based solvents, and ether-based solvents.
4. The method according to claim 2, wherein the alkali metal compounds are alkali metal salts.
5. The method according to claim 2, wherein the salts of phosphorous-containing acids are hydrates and/or alkali metal salts.
6. The method according to claim 2, wherein the salts of organic acids are hydrates and/or alkali metal salts.
7. The method according to claim 1, wherein the epoxy resin-cured product contains a halogen atom.
8. The method according to claim 1, wherein a decomposition product of the epoxy resin-cured product contains a compound recyclable as a raw material of

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synthetic resins.

9. The method according to claim 1, wherein the epoxy resin-cured product is treated under atmospheric pressure.

10. The method according to claim 1, wherein the treatment  
5 liquid has a temperature of 250°C or lower in air and 300°C or lower in an inert gas.

11. A method of separating a composite material of inorganic matter and epoxy resin-cured product, comprising the steps of:

10 (1) treating the composite material of inorganic matter and epoxy resin-cured product with a treatment liquid containing a decomposition catalyst for epoxy resin-cured products and an organic solvent to decompose and dissolve the epoxy resin-cured product; and

15 (2) separating the inorganic matter from the liquid obtained in the step (1).

12. The method according to claim 11, wherein the decomposition catalyst for epoxy resin-cured products contains one or more compounds selected from alkali metals  
20 and/or alkali metal compounds, phosphorous-containing acids and/or salts thereof, and organic salts and/or salts thereof.

13. The method according to claim 11, wherein the organic solvent contains one or more solvents selected from amide-based solvents, alcohol-based solvents, ketone-based  
25 solvents, and ether-based solvents.

14. The method according to claim 12, wherein the alkali metal compounds are alkali metal salts.





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decomposition catalyst for epoxy resin-cured products contains one or more compounds selected from alkali metals and/or alkali metal compounds, phosphorous-containing acids and/or salts thereof, and organic salts and/or salts thereof.

- 5 25. The treatment liquid according to claim 23, wherein the organic solvent contains one or more solvents selected from amide-based solvents, alcohol-based solvents, ketone-based solvents, and ether-based solvents.

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Docket #: Bernstein Dec HTK9

# Declaration and Power of Attorney for Utility or Design Patent Application

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My residence, post office address and citizenship are as stated below next to my name:

I believe I am the original, first and sole inventor (if only one name is listed below) or an original, first and joint inventor (if plural names are listed below) of the subject matter which is claimed and for which a patent is sought on the invention entitled

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私は、前記のとおり補正した請求の範囲を含む前記明細書の内容を検討し、理解したことを陳述する。

私は、連邦規則法典第 37 編第 1 条 56 項に定義されているとおり、特許資格の有無について重要な情報を開示すべき義務があることを認めます。

私は、合衆国法典第 35 部第 119 条 (a-d) 項又は第 365 条 (b) 項に基づく、下記の外国特許出願又は発明者証出願、或いは第 365 条 (a) 項に基づく、少なくとも米国以外の 1 カ国を指名した PCT 国際出願の外国優先権を主張し、更に優先権の主張に係わる基礎出願の出願日前の出願日を有する外国特許出願、又は発明者証出願或るいは PCT 国際出願を以下に“なし”の箱に印をつけることにより明記する：

Prior foreign applications  
先の外国出願

<b>P11-286187</b>	<b>Japan</b>	<b>10/Jul/99</b>
(Number)	(Country)	(Day/Month/Year Filed)
(番号)	(国名)	(出願の年月日)

<b>P11-286188</b>	<b>Japan</b>	<b>10/Jul/99</b>
(Number)	(Country)	(Day/Month/Year Filed)
(番号)	(国名)	(出願の年月日)

☐ その他の外国特許出願番号は別紙の追補優先権欄にて記載する。

☐ Additional foreign application numbers are listed a supplemental priority sheet attached hereto.

the specification of which is attached hereto unless the following box is checked:

☒ was filed on **05/Oct/00** as United States Application Number **10/088,531** and was amended on (if applicable) or,

PCT International Application Number **PCT/JP00/06951** and was amended on (if applicable).

I hereby state that I have reviewed and understand the contents of the above identified specification, including the claims, as amended by any amendment referred to above.

I acknowledge the duty to disclose information which is material to patentability as defined in Title 37, Code of Federal Regulations, §1.56.

I hereby claim foreign priority under Title 35, United States Code §119(a-d) or §365(b) of any foreign application(s) for patent or inventor's certificate, or §365(a) of any PCT international application which designated at least one country other than the United States, listed below. I have also identified below, by checking the "No" box, any foreign application for patent or inventor's certificate, or of any PCT international application having a filing date before that of the application on which priority is claimed:

Priority claimed  
優先権の主張

<input checked="" type="checkbox"/>	<input type="checkbox"/>
Yes	No
あり	なし

<input checked="" type="checkbox"/>	<input type="checkbox"/>
Yes	No
あり	なし



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**Japanese Language Utility or Design Patent Application Declaration**

委任状： 私は、下記発明者として、下記に明記された顧客番号を伴う以下の弁護士又は、代理人をここに選任し、本願の手続きを遂行すること並びにこれに関する一切の行為を特許商標庁に対して行うことを委任する。そして全ての通信はこの顧客番号宛に発送される。

顧客番号 7055

現在委任された弁護士は下記の通りである。

Neil F. Greenblum Reg. No. 28,394  
 Bruce H. Bernstein Reg. No. 29,027  
 James L. Rowland Reg. No. 32,674  
 Arnold Turk Reg. No. 33,094

POWER OF ATTORNEY: As a named inventor, I hereby appoint the attorney(s) and/or agent(s) associated with the Customer Number provided below to prosecute this application and transact all business in the Patent and Trademark Office connected therewith, and direct that all correspondence be addressed to that Customer Number:

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同第二共同発明者の署名	日付	Second Inventor's signature	<u>Hiroshi Shimizu</u> Date <u>May/13/2002</u>
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(第三またはそれ以降の共同発明者に対しても同様な情報および署名を提供すること。)

(Supply similar information and signature for third and subsequent joint inventors.)

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## Japanese Language Utility or Design Patent Application Declaration

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共同発明者の署名	日付	Fifth Inventor's signature	Date
住所		Residence	
国籍		Citizenship	
郵便の宛先		Post Office Address	
第六の共同発明者の氏名 (該当する場合)		Full name of sixth joint inventor, if any	
共同発明者の署名	日付	Sixth Inventor's signature	Date
住所		Residence	
国籍		Citizenship	
郵便の宛先		Post Office Address	

(それ以降の共同発明者に対しても同様な情報および署名を提供すること。)

(Supply similar information and signature for subsequent joint inventors.)